

Supplemental Problems Stoichiometry Answer



Supplemental Problems Stoichiometry Answer

Chemistry: Supplemental Stoichiometry Problems Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise specified. Show your work, including proper units, to earn full credit. ... ANSWERS: 2. 18.0 mol O₂ 3B. 13.8 g NaCl 3C. NaNO₃ 3D. 0.118 mol Cu(NO₃)₂ 3E. 4.2 g CuCl₂.

Name: Hour: Date: Chemistry: Supplemental Stoichiometry ...

Chemistry: Supplemental Stoichiometry Problems. Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise . specified. Show your work, including proper units, to earn full credit. 1. Given the balanced equation, show what the following molar ratios (i.e., the conversion factors ...

Stoichiometry Worksheet - teachnlearnchem.com

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Supplemental Stoichiometry Problems 1. (a) Iron (III) oxide can be reduced with carbon monoxide according to the following equation. $\text{Fe}_2\text{O}_3(\text{s}) + 3 \text{CO}(\text{g}) \rightarrow 2 \text{Fe}(\text{s}) + 3 \text{CO}_2(\text{g})$ A 16.2 L sample of CO (g) at 1.50 atm and 200. °C is combined with 15.39g of Fe₂O₃(s).

Supplemental Stoichiometry Problems

These problems are provided for each of the chapters for which additional mathematical problems would be beneficial. Most chapters contain 10–25 supplemental problems. You might use them as assessments or assign them for homework. Complete solutions can be found at the back of the Supplemental Problemsbook. To the Teacher

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Stoichiometry Worksheet #1 Answers 1. Given the following equation: $2 \text{C}_4\text{H}_{10} + 13 \text{O}_2 \rightarrow 8 \text{CO}_2 + 10 \text{H}_2\text{O}$, show what the following molar ratios should be. a. $\text{C}_4\text{H}_{10} / \text{O}_2$ b. O_2 / CO_2 c. $\text{O}_2 / \text{H}_2\text{O}$ d. $\text{C}_4\text{H}_{10} / \text{CO}_2$ e. $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$ 2. Given the following equation: $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$

2 a. How many moles of O_2 can be produced by ...

Stoichiometry Worksheet #1 Answers

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Chapter 11 Supplemental Problems Stoichiometry Answers

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Supplemental Problems Stoichiometry Answer

Supplemental Modules (Inorganic Chemistry) ... Step 3: Answer the question of what is being asked. The question asks how much $H_2(g)$ was produced. You are expected to solve for the amount of product formed. ... Problems. Stoichiometry and balanced equations make it possible to use one piece of information to calculate another. There are ...

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